

**Instructions to candidates**

**Answer all questions.**

**Duration:** Two (02) hours

**Number of questions:** Four (04)

**Point allocation:** 100 marks

1.
  - a. Write an expression for the **electronic** Hamiltonian for the carbon atom (use atomic units). (10 marks)
  - b. Classify the various terms that are included in the above Hamiltonian as electronic kinetic energy, nuclear-electronic attraction, and electron repulsion. (7 marks)
  - c. Which energy contributions are not included in the electronic Hamiltonian that are included in the full Hamiltonian? (8 marks)
2.
  - a. Explain what basis functions are as used in electronic structure calculations. (5 marks)
  - b. State the Gaussian product rule? (5 marks)
  - c. Discuss the pros and cons in the usage of Slater-Type Orbitals (STOs) and Gaussian-type orbitals (GTOs)? (7 marks)
  - d. What is the difference between STO-1G and STO-3G basis sets? (8 marks)
3.
  - a. What is meant by a potential energy surface (PES)? (5 marks)
  - b. State the Born-Oppenheimer (BO) approximation? (5 marks)
  - c. Describe the variational theorem. (5 marks)
  - d. What was the drawback of the wavefunction proposed by the Hartree approach? (5 marks)
  - e. How does the Hartree – Fock approach overcome the above-mentioned drawback? (Explain using an example) (5 marks)
4.
  - a. What is a “functional” as opposed to a “function” in mathematics? (Describe using an appropriate example) (5 marks)
  - b. Briefly describe the Kohn-Sham (KS) approach to Density Functional Theory (DFT). (5 marks)
  - c. According to the mathematical formulation of KS approach to DFT, what are the terms that combine to produce the “exchange correlation functional” ? (5 marks)
  - d. Describe the following different approaches for approximating exchange-correlation potential.
    - i. Local density approximation (LDA) (5 marks)
    - ii. Local spin density approximation (LSDA) (5 marks)

